

Rotational Isomers in the Crystals of DL- α -Amino-*n*-butyric Acid

Tadayoshi ICHIKAWA, Yoichi IITAKA and Masamichi TSUBOI

Faculty of Pharmaceutical Sciences, The University of Tokyo, Hongo, Bunkyo-ku, Tokyo

(Received October 12, 1967)

For DL- α -amino-*n*-butyric acid, two crystalline modifications were previously found.¹⁾ They were called as A- and B-forms. A-form is monoclinic, with the space group of $P2_1/a$, and with the unit cell dimensions: $a=9.85$ Å, $b=4.77$ Å, $c=11.94$ Å, $\beta=101^\circ$, $Z=4$. B-form is tetragonal, with $P4_2/n$, $a=13.40$ Å, $c=5.82$ Å, and $Z=8$. By our recent X-ray analyses, it has now been established that all of the three rotational isomers (I, II, and III) shown in Fig. 1 are coexisting in the A-form crystal, whereas only one conformation (II in Fig. 1) is found in the B-form crystal.

Three-dimensional intensity data were collected from multiple-film equi-inclination Weissenberg photographs taken with filtered $\text{CuK}\alpha$ radiation. For the A-form crystal, data of 809 reflections were used. First, the sharpened three-dimensional Patterson function was calculated. From this function, approximate coordinates of two oxygen atoms were estimated. The signs based upon these coordinates were used in the first three-dimensional Fourier synthesis. The coordinates of all the heavy atoms except those of the γ -carbon could be obtained in the Fourier map. In the course of the refinement, the γ -carbon atom was found to be distributed over the three positions corresponding to the three rotational isomers shown in Fig. 1. Therefore, the site occupancy factors of the γ -C in the three positions were set as para-

meters in the least squares refinement. After two cycles of full matrix least squares calculation with anisotropic temperature factors, the R factor dropped to 0.13. At the right end of Fig. 1, an electron density map of a section through the centers of the three possible methyl groups is given, which is obtained from the difference Fourier synthesis. This map shows how the methyl group (and two hydrogen atoms) on the β -carbon are distributed among the possible sites in the internal rotation around the $\text{C}_\alpha\text{-C}_\beta$ axis. The dihedral angles $\text{C}_7\text{C}_\beta\text{C}_\alpha\text{N}$ determined are given in Table 1.

TABLE 1. DIHEDRAL ANGLES^{a)} $\text{C}_7\text{C}_\beta\text{C}_\alpha\text{N}$ IN α -AMINO-*n*-BUTYRIC ACID

A - form	I	72.4°	(36% ^{b)})
	II	-174.6	(40%)
	III	-53.8	(24%)
B - form		-176.0	

a) This angle is defined so that it is zero when $\text{C}_\beta\text{-C}_7$ bond comes at the eclipsed position with respect to the $\text{C}_\alpha\text{-N}$ bond. The positive direction is defined as that of the anticlockwise rotation of the $\text{C}_\beta\text{-C}_7$ bond, when viewed along the $\text{C}_\beta\text{-C}_\alpha$ bond as shown in Fig. 1.

b) Normalized site occupancy factors as determined by the least squares calculation.

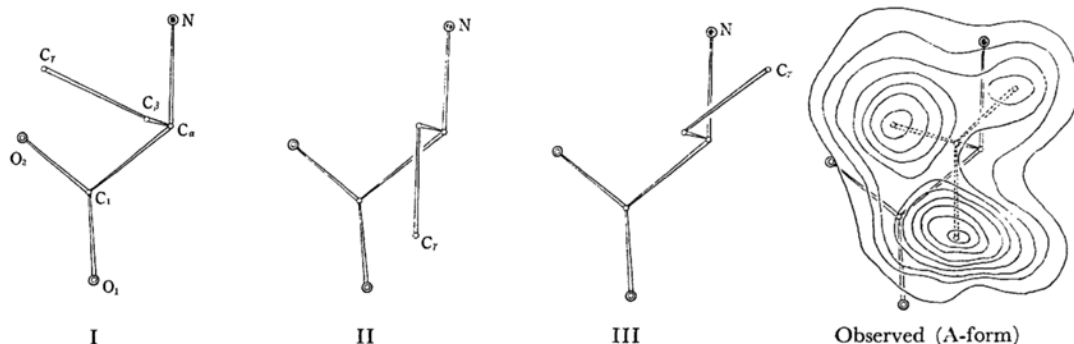


Fig. 1. Rotational isomers I, II, and III expected for α -amino-*n*-butyric acid. What is given at the right end is a section of the electron density map at the CH_3 group of A-form. This was obtained by a difference Fourier synthesis by the use of the structure factors calculated on the basis of all the heavy atoms except the γ -carbon. In this synthesis, there appeared three maxima corresponding to the three stable positions of C_7 , and this section was prepared through the plane determined by these three points. The figure is the projection along the c -axis of this section. Contours are drawn at 0.1, 0.2, ..., 0.9 electron Å⁻³.

1) M. Tsuboi, Y. Iitaka, S. Suzuki and S. Mizushima, *This Bulletin*, **32**, 529 (1959).

For the B-form crystal, intensities of 630 reflections were measured. The Harker section at $w=1/2$ allowed us to estimate only two-dimensional (x and y) coordinates of oxygen atoms and it was not possible to estimate the z parameters from the Patterson function. Accordingly, the analysis was started with two dimensional Fourier methods. After the x and y coordinates of all the heavy atoms (except C_7) were determined, the approximate z coordinates of these atoms were estimated on the basis of a model-building. The first three-dimensional Fourier synthesis was made on the basis of the estimated coordinates of the two oxygen and one nitrogen atoms. All the carbon atoms including C_7 appeared at reasonable

positions. C_7 appeared at the trans position to the NH_3^+ group, with the dihedral angle of -176° (II of Fig. 1). It was ascertained that C_7 takes only this site by a three-dimensional difference Fourier synthesis based upon the coordinates of the heavy atoms except C_7 . After three cycles of least squares refinement with isotropic temperature factors, the R value reached 0.12.

More detailed crystallographic data as well as a full account of the crystal structures of the A- and B-forms of this amino acid will be given elsewhere.

This work was supported partly by a grant from the Ministry of Education, and partly by a grant from the United States Public Health Service GM 10024-04.
